

## PhD offer

### Project title: Non-noble metal catalysts with tetra-aza-derived macrocyclic ligands for electrocatalytic reduction of protons and CO<sub>2</sub> (Tetra-azaCat)

**Starting – Duration :** October 2024 – 36 months

**Location:** Département de Chimie Moléculaire (DCM) UMR 5250 UGA-CNRS  
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**Project description:** The development of artificial photosynthesis systems for water-splitting and CO<sub>2</sub> reduction for hydrogen and valuable chemicals production (CO, HCOOH...) requires the development of efficient, selective and stable catalysts capable of operating at low overpotential and based on earth abundant transition metals. Cobalt complexes with tetra- and penta-aza macrocyclic ligands including the pyridyldiimine motif constitute a very promising family of catalysts recently exploited for the electro- and photo-induced reduction of protons to hydrogen (HER) and the reduction of CO<sub>2</sub> (CO<sub>2</sub>RR). In particular, the cobalt tetra-aza complex has received an increase interest since it is now considered as one of the most efficient and stable H<sub>2</sub>-evolving catalyst (HEC) in fully aqueous solution. In addition, it has been recently shown that a simple substitution of the pyridine moiety of the macrocycle ligand by a pyridine or pyridinium allows selectivity for CO<sub>2</sub>RR to be directed toward CO formation over H<sub>2</sub>.

The aim of this thesis project is the development of this family of catalysts, still under-exploited by introducing various substituents on the macrocycle ligand. These substituents could induce a modification of the electronic properties of the metal center, act as axial ligands and/or proton relays and thus modify/enhance the reactivity, by promoting electron transfer coupled to proton transfer (PCET) reactions and/or directing the selectivity for HER or CO<sub>2</sub>RR. As the complexes are synthesized by template effect, i.e. by reaction of the metal cation with bis-acetyl pyridine and a polyamine, precursors bearing different substituents will first be synthesized. The structures of all new complexes will be ascertained by X-ray diffraction. Electrochemical and spectroscopic properties (NIR/Vis. and EPR) of the initial complexes and their electrogenerated reduced forms as well as their performance as electrocatalysts for HER and CO<sub>2</sub>RR in terms of efficiency, selectivity (for CO<sub>2</sub>RR) and robustness will be assessed by cyclic voltammetry and electrolysis under homogeneous conditions. This project aims also to establish correlations between the catalytic performance, and the selectivity of the new complexes with tetra-aza derived macrocyclic ligands with their structural, electronic/spectroscopic properties.

This PhD thesis will be carried out in collaboration between two teams of the DCM, the Electrochimie Moléculaire et Photochimie Redox (EMPre) team and the Synthèse et réactivité en Chimie Organique (SeRCO) team.

**Keywords :** cobalt catalyst, macrocyclic ligand, coordination chemistry, molecular electrochemistry, electrocatalysis, small molecule activation, artificial photosynthesis.

**Applicant profile:** The candidate should hold a master's degree in chemistry with interest for synthesis, coordination chemistry and molecular electrochemistry. Experience in electrocatalysis would be an asset.

**Funding:** Labex ARCANE, CBH Graduate School

**Application:** Apply before April 8<sup>th</sup>, 2024 (CV and motivation letter)

## Selected references

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